

Introduction

- Accidental internal contamination of workers with highly radiant submicronic cobalt oxide particles ($^{60}\text{Co}_3\text{O}_4\text{P}$) may occur during procedures such as maintenance
- Due to their poor solubility, $^{60}\text{Co}_3\text{O}_4\text{P}$ may be retained in the lung macrophages for long periods of time, potentially causing pulmonary damage
- Soluble Co species are rapidly eliminated after intake by urinary excretion
- Calcium-trisodium diethylenetriamine-pentaacetic acid (Ca-DTPA) is the recommended treatment following cobalt intake although no proof of efficacy has been reported following inhalation of $^{60}\text{Co}_3\text{O}_4\text{P}$
- By combining the reductant ascorbic acid, to alter the redox state of Co and destabilize the surface of the $\text{Co}_3\text{O}_4\text{P}$, with a Co chelating agent (DTPA), we aim at increasing the dissolution of $\text{Co}_3\text{O}_4\text{P}$ and thus enhance urinary excretion following accidental intake

- Aims:**
- To better understand the relationship between physicochemical properties and biological behavior of $\text{Co}_3\text{O}_4\text{P}$
 - To provide therapeutic approaches in case of accidental intake

Methods

Particles : Co_3O_4 (Sigma) mean diameter 372 ± 101 nm

Activation by exposure to external neutron beam

« ISIS »: fast neutrons 500 kW

« ILL »: pure cold neutrons 42 MW

Resulting activated particles contain stable ^{59}Co plus radioactive ^{60}Co and are referred to as ^{60}Co / $\text{Co}_3\text{O}_4\text{P}$

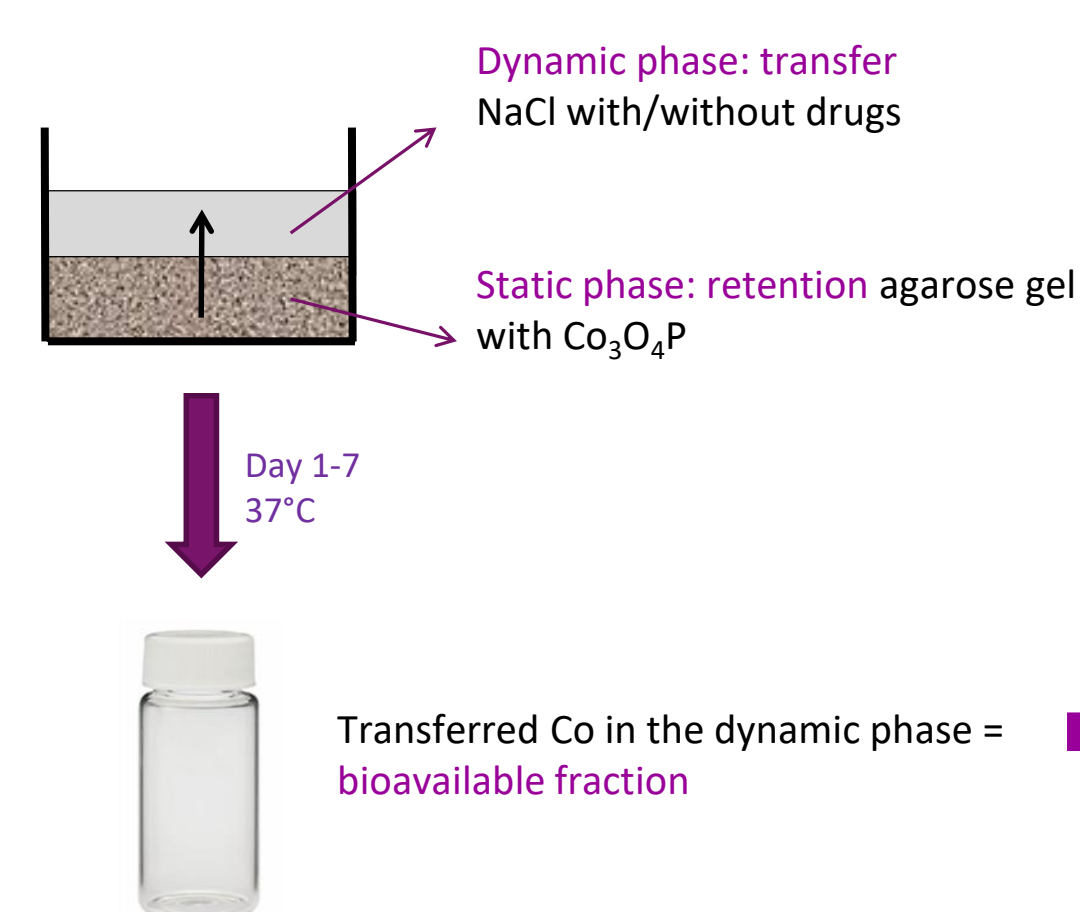
	kBq/mg	^{60}Co /total Co
ISIS particles	12	3×10^{-7}
ILL particles	25	6×10^{-7}

In vitro models provide information regarding

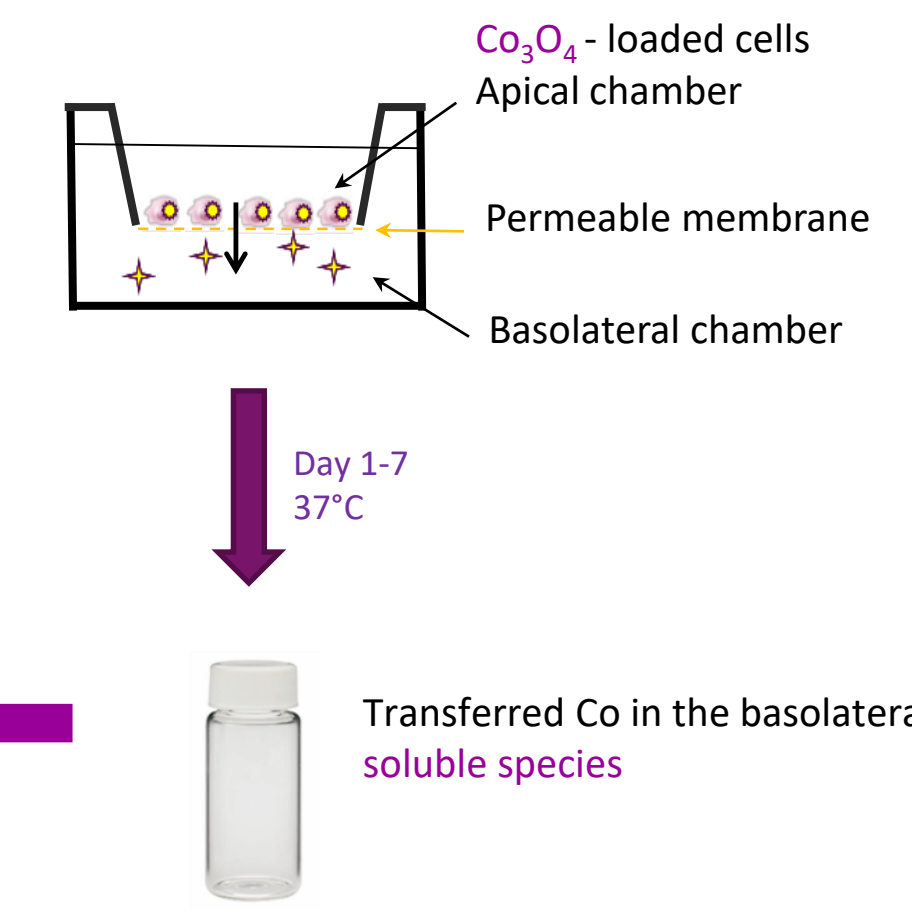
- Intrinsic dissolution properties of $\text{Co}_3\text{O}_4\text{P}$ according to their activation status

- Activation-induced changes of surface properties (Ascorbate) and availability to chelation (DTPA)

Acellular model: Transfer from a static phase (retention compartment) to a dynamic phase (transfer compartment)



Cellular model: Dissolution of particles following phagocytosis by macrophage-like cells (THP-1)



Cobalt/activity measurement:

ICP-MS for Co quantification or gamma spectrometry (Germanium detector) for activity measurement

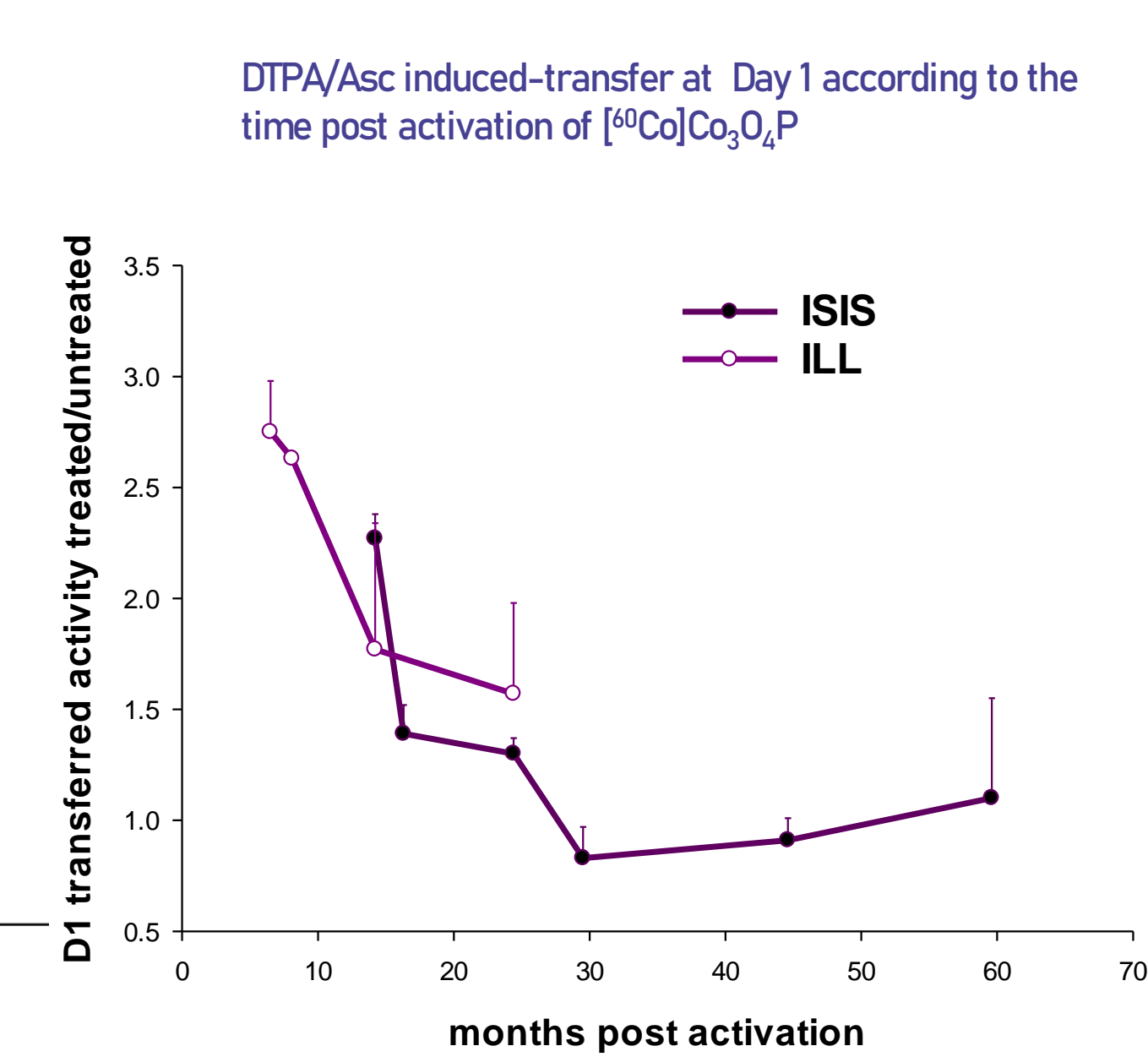
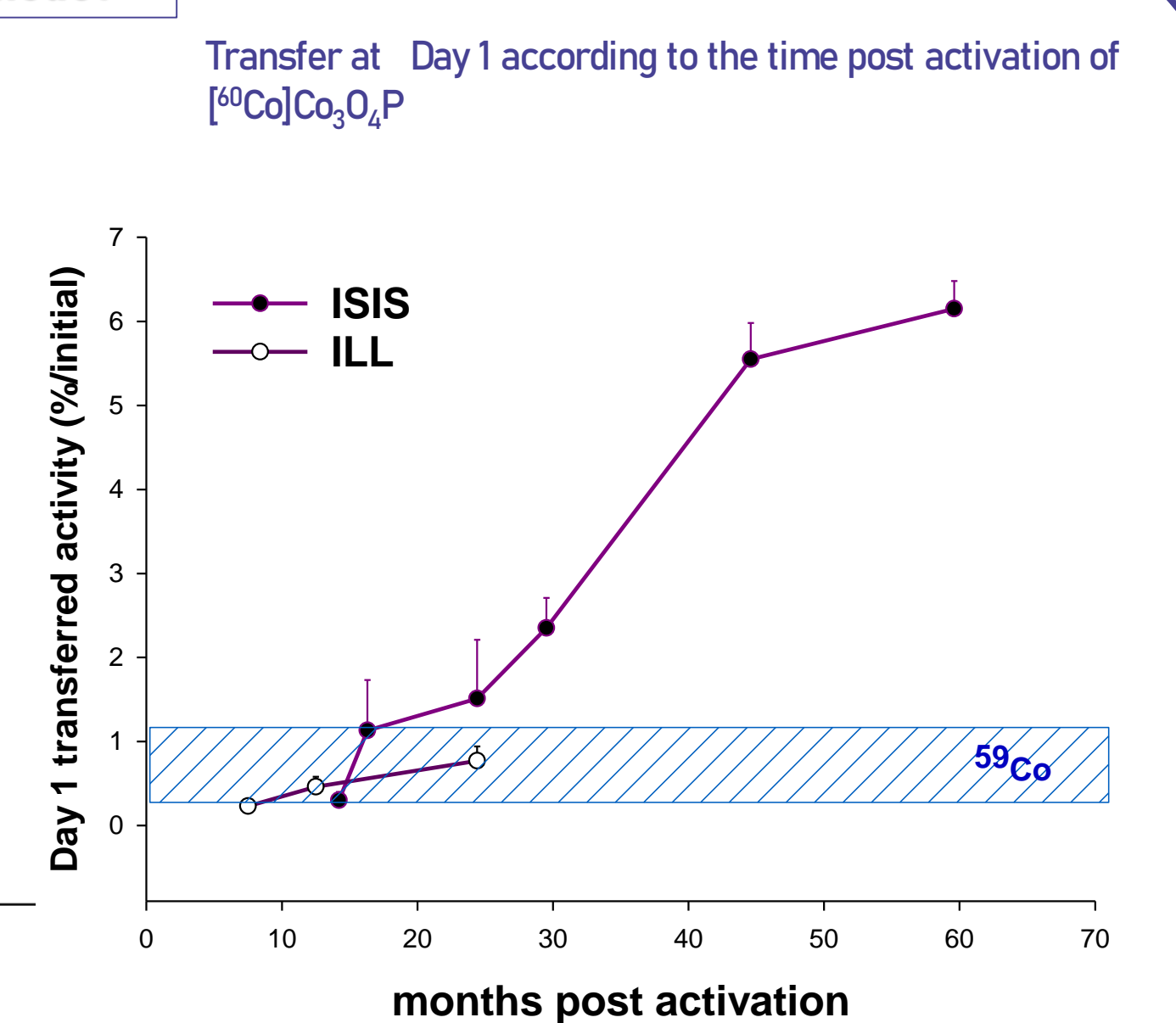
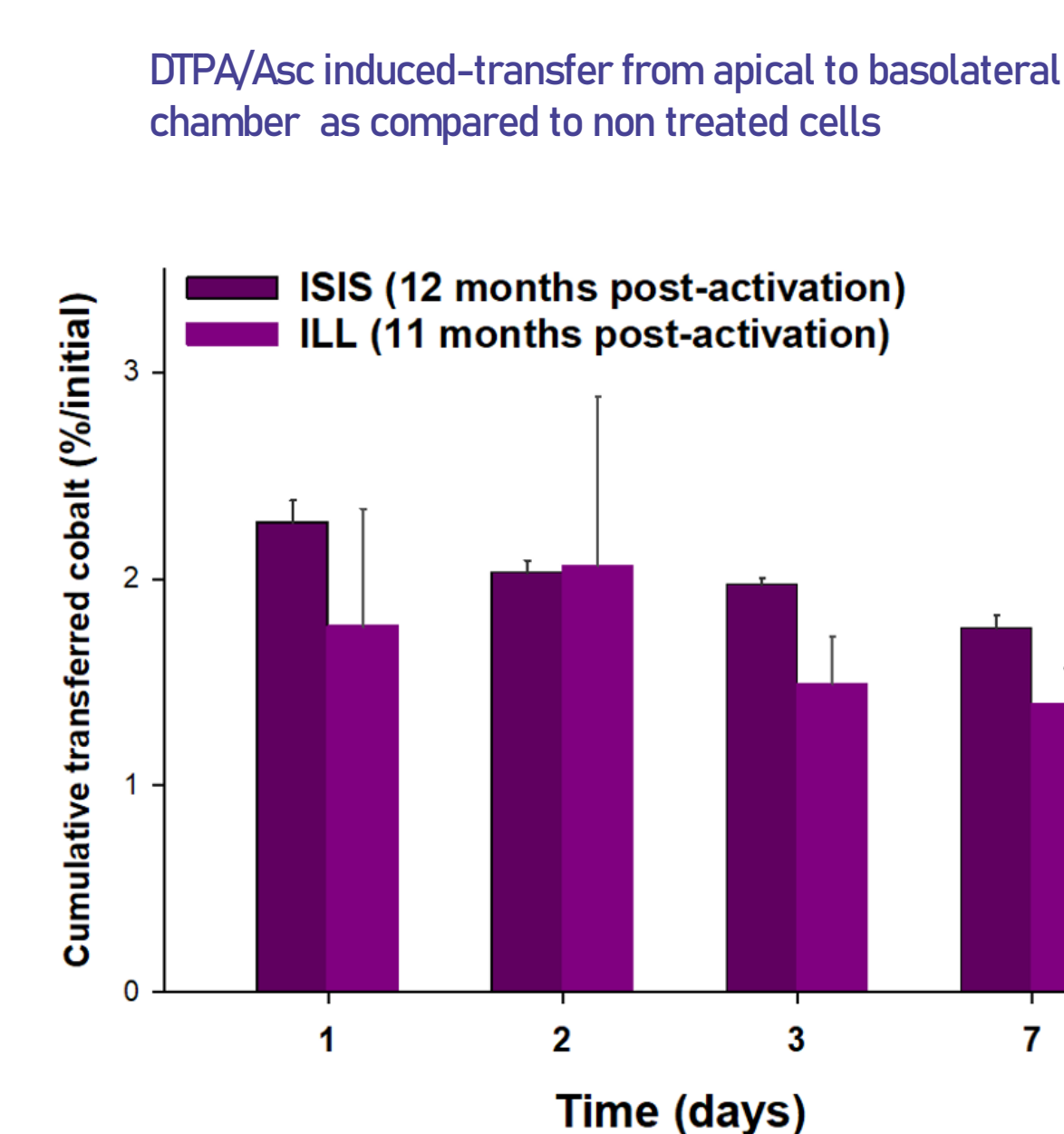
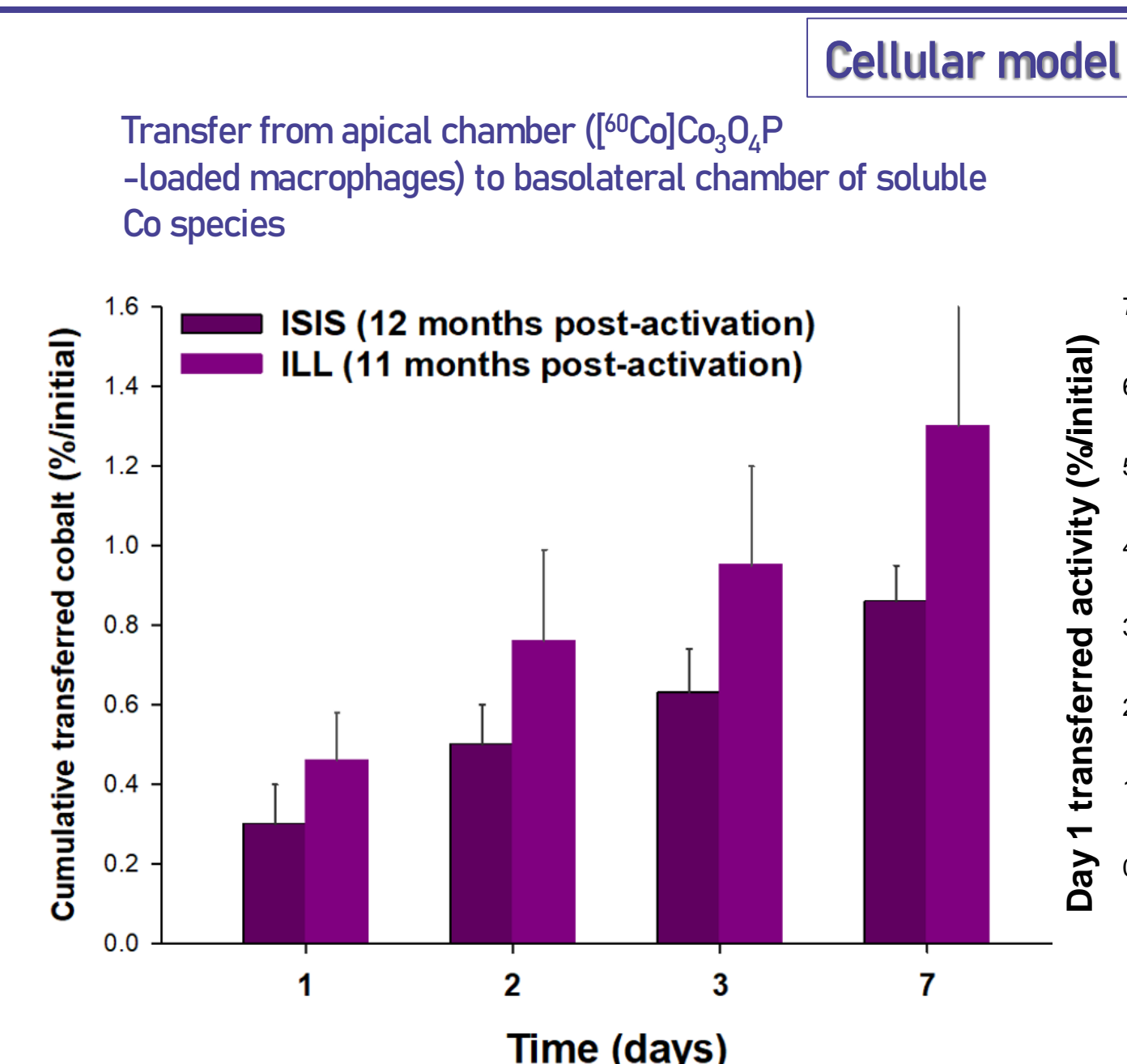
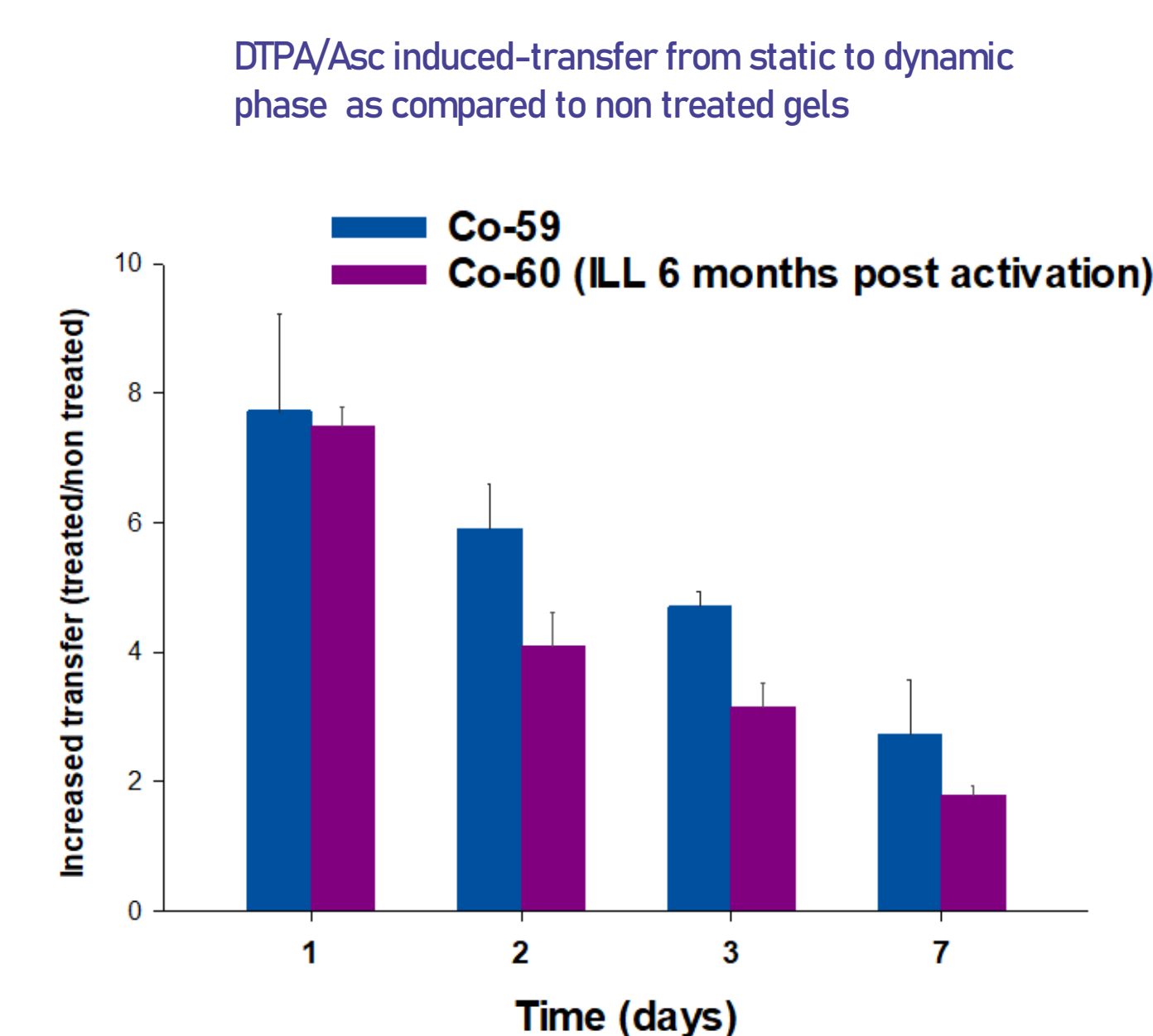
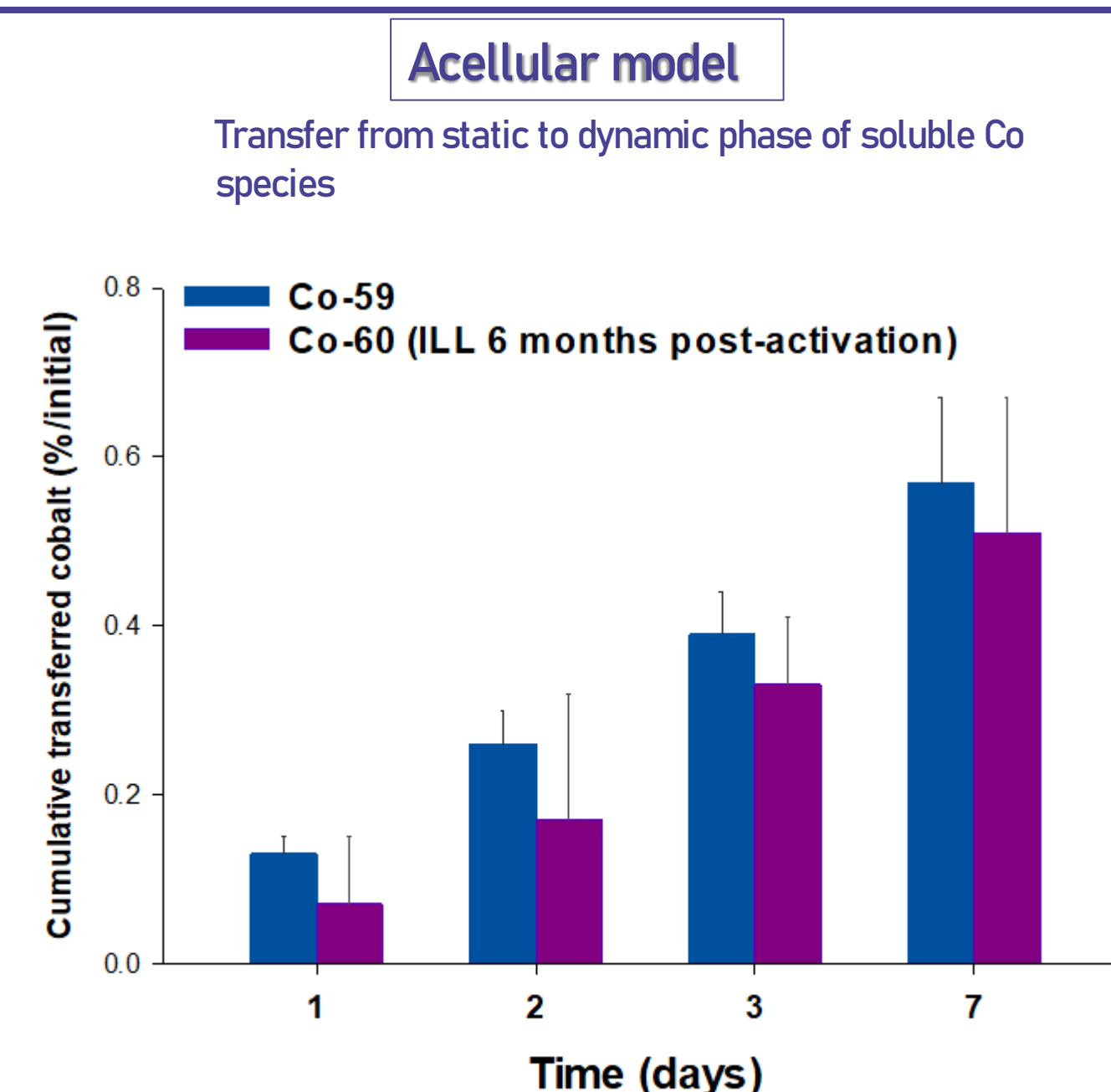
Results

Intrinsic dissolution properties and effect of aging

- Early after activation (within a year), dissolution properties of ^{60}Co / $\text{Co}_3\text{O}_4\text{P}$ evaluated with acellular and cellular models are similar to that of stable particles
- Dissolution from macrophages increases with time post activation

Effect of treatments

- DTPA/Asc increases the dissolution/bioavailability of $\text{Co}_3\text{O}_4\text{P}$ with a similar effect whatever the activation status of the particles
- In acellular model, the enhanced transfer in the presence of DTPA/Asc decreases with the time after incubation whereas in cellular models, the efficacy of treatment is stable over the time of THP-1 incubation
- The effect of treatment decreases slightly depending on activation conditions (ISIS versus ILL) and the time post activation



Conclusions

- Neutron activation seems to modify the dissolution/bioavailability of Co_3O_4 particles with a time-dependent post-activation effect
- DTPA/Asc increases the dissolution/bioavailability of Co_3O_4 particles but the extent to which the treatment is effective may depend on the initial status of the particles (activation and time post-activation)
- In vivo* rat studies are necessary to evaluate decorporation efficacy of DTPA/Asc following pulmonary intake according to the physicochemical properties of $\text{Co}_3\text{O}_4\text{P}$

References

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